

- 1 (a) SOLID higher temperature means higher energy/greater speed of mols/particles/atoms NOT more vibration NOT vibrate more B1
- vibrations get bigger or movement greater/take up more space or separation larger B1
- GAS (ave) speed/energy of mols/particles/atoms greater B1
- (ave) separation of mols/particles/atoms greater or mols/particles/atoms take up more space or increased pressure causes container to get bigger B1
- (b) liquids: slightly more B1
- gases: much more B1
- (c) regular/uniform expansion or appropriate range (be generous if numbers quoted) or expands a lot/large expansivity or (relatively) non-toxic or low freezing point/melting point or measures low temperatures any 1 B1
- IGNORE reacts to small temp change IGNORE high boiling point [7]
- 2 (a) (i) funnel no longer giving heat to ice OR ice at M.P./constant temp OR heater reached max temp B1
- (ii) inside of large pieces could be well below freezing point)
OR smaller air gaps if pieces smaller) any 1 B1
OR better contact between heater and ice)
OR to ensure heat from heater only goes to the ice)
OR larger surface area)
Ignore ice melts faster
- (b) mass of beaker NOT mass of ice NOT mass of water B1
mass of beaker + water B1
(apply $\checkmark + \times = 0$ for extras other than power & time)
- (c) (mass of ice melted by heater = $16.3 - 2.1$) = 14.2 g C
ml in any form, words, symbols or numbers C1
Wt or Pt in any form, words, symbols or numbers accept VIit C
338 J/g OR 338 000 J/kg c.a.o A1 [8]

- 3 (a) total mass before ice added B1
total mass after all ice melted B1 [2]
- (b) (i) mass \times sp ht cap \times change in temp or 20 OR $mc\theta$ B1 [1]
(ii) mass (of melted ice) \times sp latent ht OR ml
OR (heat gained by ice) = heat lost by water B1 [1]
- (c) heat/mass or 12 800/30 C1
427 J/g OR 426667 J/kg any no s.f. ≥ 2 A1 [2]
- (d) heat gained from surroundings OR no lagging B1
heat needed to cool beaker/stirrer and thermometer) any 2 +
too much ice added or similar point) B1 [2]
allow stirring gives energy, allow evaporation/condensation
(ignore "mistakes when taking readings" or similar)

[Total: 8]

- 4 (a) on surface/throughout; no bubbles/bubbles; all temps./b.p.;
s.v.p. < at. pressure; svp = at. pressu any two B2 2
- (b) energy/work to separate molecules B1
(against) forces of attraction between water molecules B1 2
(to break bonds C1)
The k.e./speed of the molecules does not increase B1 1
- (c) Wt = mL or 120 x 1 = 0.05 x L C1
L = 120/0.05 C1
L = 2400 J/g A1 3

[8]

- | | | | |
|---|---|------------------|--------------------------------|
| 5 | <p>(a) increase surface area of tank
blow air over surface/put in windy place</p> | <p>B1
B1</p> | <p>2</p> |
| | <p>(b) (i) capillary tube longer or liquid with lower expansivity</p> | <p>B1</p> | |
| | <p>(ii) capillary tube thinner/finer or liquid with higher expansivity
or bigger bulb</p> | <p>B1</p> | <p>2</p> |
| | <p>(c) $p_1v_1 = p_2v_2$ or $1 \times 10^5 \times 150 = p_2 \times 50$
$p_2 = 3 \times 10^5$ (Pa)</p> | <p>C1
A1</p> | <p>2
[6]</p> |